_{eacher:} Finnegan		Finnegan	subject: Science		Course: AP Cher	n Grade: <u>1</u>	Date(s) : ^{8/12-8/16}	
Stai	ndaro	^{d:} Unit 1: Atomic Struc	cture and Properties					
Ass	essm	ient: 🗆 Quiz 🗆	Unit Test	Project	🖬 Lab	□ None		
	\$	Pre-Teaching	Activation of Learning (5 min)	Focused Instruction (10 min) *I DO	Guided Instruction (10 min) *WE DO	Collaborative Learning (10 min) *Y'ALL DO	Independent Learning (10 min) *YOU DO	Closing (5 min)
	() () () () () () () () () () () () () (Learning Target Success Criteria 1 Success Criteria 2	 Do Now Quick Write* Think/Pair/Share Polls Notice/Wonder Number Talks Engaging Video Open-Ended Question 	 Think Aloud Visuals Demonstration Analogies* Worked Examples Nearpod Activity Mnemonic Devices* 	 Socratic Seminar * Call/Response Probing Questions Graphic Organizer Nearpod Activity Digital Whiteboard 	 Jigsaw* Discussions* Expert Groups Labs Stations Think/Pair/Share Create Visuals Gallery Walk 	 Written Response* Digital Portfolio Presentation Canvas Assignment Choice Board Independent Project Portfolio 	 Group Discussion Exit Ticket 3-2-1 Parking Lot Journaling* Nearpod
Monday	0	I am learning about atomic structure, stoichiometry, and isotopes.	Sci. Not. Question.	Teacher led note taking and examples.	Teacher led note taking and examples.	Student pairs or single work on examples.	Student pairs or single work on examples.	Check progress.
		I can draw models of Bohr atoms,isotopes, and ions.						
		I can read a mass spec.						
Tuesday	0	I am learning about atomic structure, stoichiometry, and isotopes.	Molar mass question.	Teacher led note taking and examples.	Teacher led note taking and examples.	Student pairs or single work on examples.	Student pairs or single work on examples.	Check accuracy of avg. atomic mass answer
	~	i can calculate percent composition.						
		I can calculate average atomic mass from isotopic mass.						
Wednesday	0	I am learning about stoichiometry and dimensional analysis.	Specific atom numbers in given moles of a substance math question.	Long multi step dimentional analysis question from teacher.	Teacher guided, student led multistep stoich conversion.	White boarding session answering various stoichiometry questions.	White boarding session answering various stoichiometry questions.	Check answers.
		I can do multistep stoichiometry problems						
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Thursday	0	I am learning about percent composition and compound complexes.	Isotope/ion notation question.	Teacher introduction to lab and possible safety hazards.	Teacher introduction to lab and possible safety hazards.	Students perform dehydration reaction.	Students perform dehydration reaction.	Check for reaction completion. Place samples in desicator.
		I can perform a dehydration reaction.						
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Friday	0	I am learning about percent composition and compound complexes.	Percent composition question.	Teacher exemplar of percent composition calculation and percent error calculation.	Teacher exemplar of percent composition calculation and percent error calculation.	Students measure final product in groups.	Students do all math calculations individually.	Turn in final answers and post lab.
		I can calculate percent composition using data collected in lab.						
		I can calculate percent error based on my results.						

*key literacy strategies